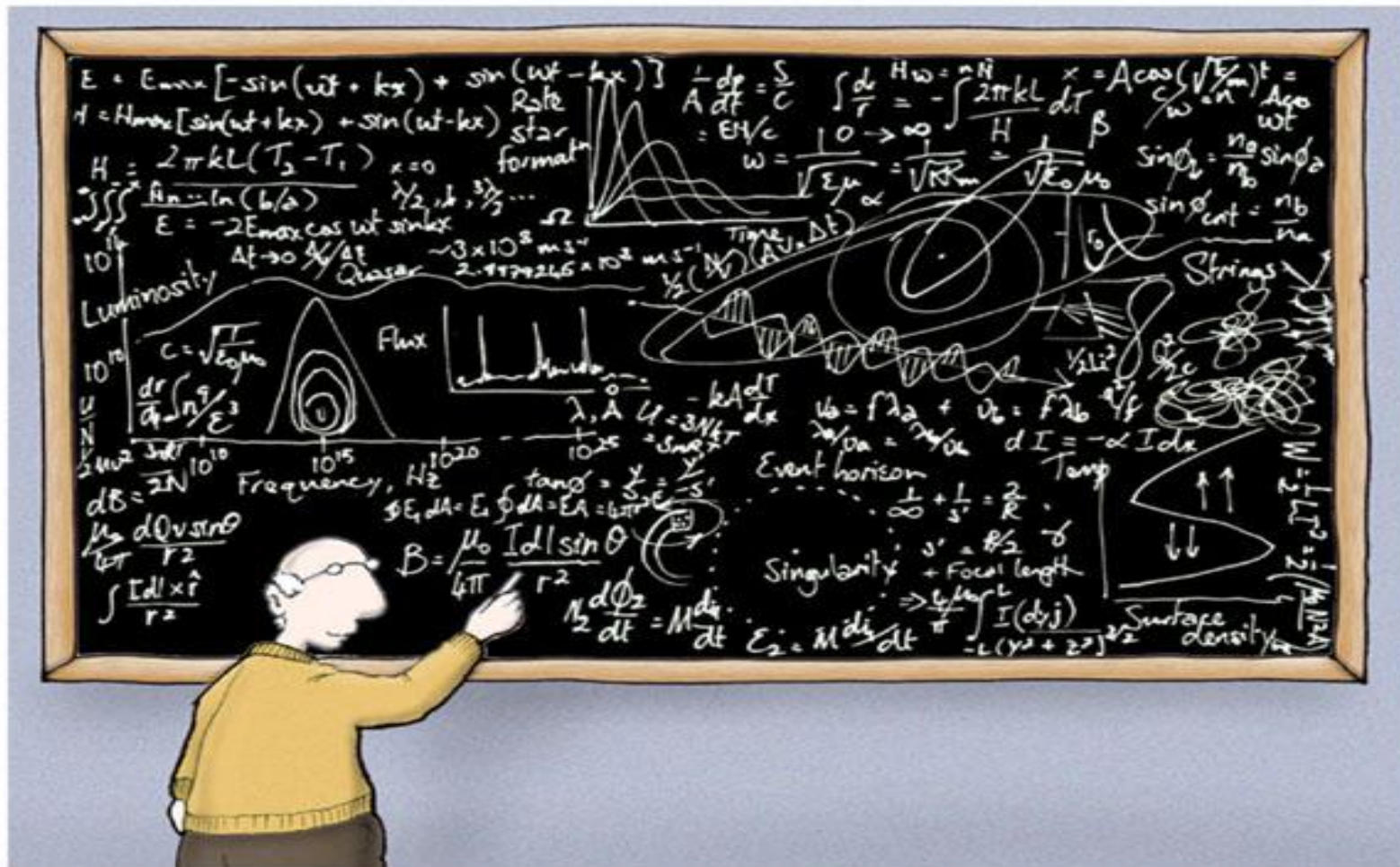


Uncertainty and Significant Figures



Astrophysics made simple

Uncertainty in Measurement

A digit that must be **estimated** is called **uncertain**. A **measurement** always has some degree of uncertainty.

Why Is there Uncertainty?

- ❖ Measurements are performed with instruments
- ❖ No instrument can read to an infinite number of decimal places

Which of these balances has the greatest uncertainty in measurement?



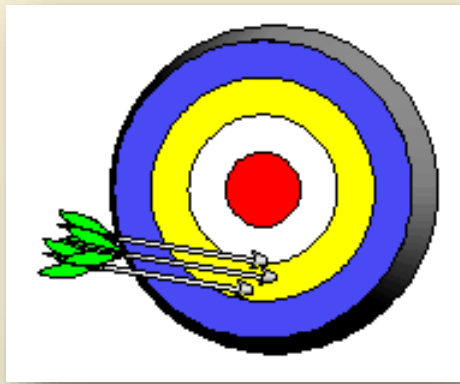
Precision and Accuracy

Accuracy refers to the agreement of a particular value with the true value.

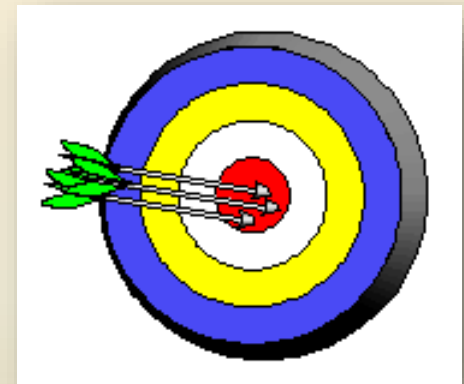
Precision refers to the degree of agreement among several measurements made in the same manner.



Neither
accurate nor
precise



Precise but not
accurate



Precise AND
accurate

Types of Error

Random Error (Indeterminate Error) - measurement has an equal probability of being high or low.

Systematic Error (Determinate Error) - Occurs in the same direction each time (high or low), often resulting from poor technique or incorrect calibration.

Rules for Counting Significant Figures - Details

Nonzero integers always count as significant figures.

3456 has
4 significant figures

Rules for Counting Significant Figures - Details

Zeros

- Leading zeros do not count as significant figures.

0.0486 has
3 significant figures

Rules for Counting Significant Figures - Details

Zeros

- **Captive zeros** always count as significant figures.

16.07 has
4 significant figures

Rules for Counting Significant Figures - Details

Zeros

A decimal to the right of **trailing zeros** makes zeros significant.

9.300 has
4 significant figures

Rules for Counting Significant Figures - Details

Zeros

Trailing zeros are significant only if the number contains a decimal point.

220. has
3 significant figures

Rules for Counting Significant Figures - Details

Exact numbers have an *infinite* number of significant figures.

1 inch = **2.54** cm, exactly

Sig Fig Practice #1

How many significant figures in each of the following?

1.0070 m → 5 sig figs

17.10 kg → 4 sig figs

100,890 L → 5 sig figs

3.29 × 10³ s → 3 sig figs

0.0054 cm → 2 sig figs

3,200,000 → 2 sig figs

Rules for Significant Figures in Mathematical Operations

Multiplication and Division: # sig figs in the result equals the number in the least precise measurement used in the calculation.

$$6.38 \times 2.0 =$$
$$12.76 \rightarrow 13 \text{ (2 sig figs)}$$

Sig Fig Practice #2

<u>Calculation</u>	<u>Calculator says:</u>	<u>Answer</u>
$3.24 \text{ m} \times 7.0 \text{ m}$	22.68 m^2	23 m^2
$100.0 \text{ g} \div 23.7 \text{ cm}^3$	$4.219409283 \text{ g/cm}^3$	4.22 g/cm^3
$0.02 \text{ cm} \times 2.371 \text{ cm}$	0.04742 cm^2	0.05 cm^2
$710 \text{ m} \div 3.0 \text{ s}$	236.6666667 m/s	240 m/s
$1818.2 \text{ lb} \times 3.23 \text{ ft}$	$5872.786 \text{ lb}\cdot\text{ft}$	$5870 \text{ lb}\cdot\text{ft}$
$1.030 \text{ g} \div 2.87 \text{ mL}$	2.9561 g/mL	2.96 g/mL

Rules for Significant Figures in Mathematical Operations

Addition and Subtraction: The number of decimal places in the result equals the number of decimal places in the least precise measurement.

$$6.8 + 11.934 = 18.734 \rightarrow 18.7 \text{ (3 sig figs)}$$

Sig Fig Practice #3

<u>Calculation</u>	<u>Calculator says:</u>	<u>Answer</u>
$3.24 \text{ m} + 7.0 \text{ m}$	10.24 m	10.2 m
$100.0 \text{ g} - 23.73 \text{ g}$	76.27 g	76.3 g
$0.02 \text{ cm} + 2.371 \text{ cm}$	2.391 cm	2.39 cm
$713.1 \text{ L} - 3.872 \text{ L}$	709.228 L	709.2 L
$1818.2 \text{ lb} + 3.37 \text{ lb}$	1821.57 lb	1821.6 lb
$2.030 \text{ mL} - 1.870 \text{ mL}$	0.16 mL	0.160 mL